**Chemical Reaction LAB DEMOS Name: \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_**

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| **Reaction Type:** | **Word Equation and****Balanced Chemical Equation** |
| **Reaction 1**type: \_\_\_\_\_\_\_\_\_  endo or exo?  | methane + oxygen gas 🡪 carbon dioxide + water \_\_\_\_\_\_\_\_\_ CH4 + \_\_\_\_\_\_\_\_\_ 🡪 \_\_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_\_\_ |
| **Reaction 2**type: \_\_\_\_\_\_\_\_\_ endo or exo? | magnesium + oxygen gas 🡪 magnesium oxide \_\_\_\_\_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_ 🡪 \_\_\_\_\_\_\_\_ |
| **Reaction 3**type: \_\_\_\_\_\_\_\_\_ endo or exo? |  hydrogen peroxide 🡪 oxygen gas + water \_\_\_\_ H2O2 🡪 \_\_\_\_\_\_\_ + \_\_\_\_\_\_ |
| **Reaction 4**type: \_\_\_\_\_\_\_\_\_ endo or exo? | potassium iodide + lead (II) nitrate 🡪 potassium nitrate + lead (II) iodide \_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_\_\_ 🡪 \_\_\_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_\_ |
| **Reaction 5**Type: \_\_\_\_\_\_\_\_\_ endo or exo? | Aluminum + Copper(II)Chloride 🡪 Aluminum Chloride + Copper \_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_\_\_ 🡪 \_\_\_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_\_ |
| **Reaction 6**type: \_\_\_\_\_\_\_\_\_ endo or exo? | Magnesium hydroxide + acetic acid 🡪 water + magnesium acetate \_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_\_\_ 🡪 \_\_\_\_\_\_\_\_\_\_\_ + \_\_\_\_\_\_\_\_\_\_ |