How Many of Each Atom?

1. 4 MgCl2

Mg\_\_\_\_\_ Cl\_\_\_\_\_

1. Be(NO3)2

Be\_\_\_\_\_ O\_\_\_\_\_ N\_\_\_\_\_\_

1. 4 Be(NO3)2

Be\_\_\_\_\_ O\_\_\_\_\_ N\_\_\_\_\_\_

1. MgCl2 + Li2CO3

Mg\_\_\_\_\_ Cl\_\_\_\_\_ Li\_\_\_\_ C\_\_\_\_\_ O\_\_\_\_

1. C6H12 O6 + 9 O2

C\_\_\_ O\_\_\_\_ H\_\_\_\_\_\_

1. 6 CO2 + 6 H2O

C\_\_\_ O\_\_\_\_ H\_\_\_\_\_\_

1. Pb + FeSO4

Pb\_\_\_ O\_\_\_\_ S\_\_\_\_\_\_ Fe\_\_\_\_\_\_

1. CaO + CO2

C\_\_\_ O\_\_\_\_ Ca\_\_\_\_\_\_

1. P4 + 3 O2

P\_\_\_\_ O\_\_\_\_\_

1. 2 P2O3

P\_\_\_\_ O\_\_\_\_\_

1. 2 RbNO3 + BeF2

Rb\_\_\_\_ O\_\_\_\_\_ N\_\_\_\_\_\_ Be\_\_\_\_\_\_ F\_\_\_\_\_\_

1. Be(NO3)2 + 2 RbF

Rb\_\_\_\_ O\_\_\_\_\_ N\_\_\_\_\_\_ Be\_\_\_\_\_\_ F\_\_\_\_\_\_

1. 2 AgNO3 + Cu

Ag\_\_\_\_ O\_\_\_\_\_ N\_\_\_\_\_\_ Cu\_\_\_\_\_\_

1. Cu(NO3)2 + 2 Ag

Ag\_\_\_\_ O\_\_\_\_\_ N\_\_\_\_\_\_ Cu\_\_\_\_\_\_

1. C3H6O + 4 O2

C\_\_\_\_ O\_\_\_\_\_ H\_\_\_\_\_\_

1. 3 CO2 + 3 H2O

C\_\_\_\_ O\_\_\_\_\_ H\_\_\_\_\_\_

1. 2 C5H5 + H2O

C\_\_\_\_ O\_\_\_\_\_ H\_\_\_\_\_\_

1. Fe(C5H5)2

Fe\_\_\_\_ C\_\_\_\_\_ H\_\_\_\_\_\_